## Highly diastereoselective additions of methoxyallene and acetylenes to chiral α-keto amides<sup>†</sup>

## So Won Youn, Yong Hae Kim,\* Jeong-Wook Hwang and Youngkyu Do\*

Center for Molecular Design & Synthesis, School of Molecular Science, Department of Chemistry, Korea Advanced Institute of Science and Technology, Taejon, 305-701, Korea. E-mail: kimyh@mail.kaist.ac.kr

Received (in Cambridge, UK) 9th January 2001, Accepted 24th April 2001 First published as an Advance Article on the web 15th May 2001

 $\alpha$ -Keto amides bearing (S)-indoline chiral auxiliaries reacted with lithiated methoxyallene or lithium acetylides to produce the corresponding dihydrofuranones 7 through formation of the tertiary  $\alpha$ -hydroxy allenes, or tertiary  $\alpha$ -hydroxy acetylenes, respectively, at -78 °C with high diastereoselectivities (up to >99% de).

A number of diastereoselective nucleophilic additions of organometallic reagents to  $\alpha$ -keto amides<sup>1</sup> bearing appropriate chiral auxiliaries have been reported as useful methods for the synthesis of optically active tertiary  $\alpha$ -hydroxy acid derivatives, which are valuable for the asymmetric syntheses of medicinal agents and natural products.<sup>2</sup> Creating a tertiary alcohol center in which the stereochemistry can be controlled by a defined chiral environment in the addition of organometallic reagents to ketones still represents a significant challenge.3

Lithiated methoxyallene<sup>4-6</sup> is a promising nucleophile because the products produced by its addition to carbonyl compounds<sup>7-8</sup> can be converted into a variety of interesting compounds such as enones9 or dihydrofuran derivatives.<sup>10</sup> In particular chiral propargylic alcohols<sup>11</sup> are useful intermediates in the synthesis of natural products.<sup>2,12</sup> Although a number of stereoselective additions of acetylides to aldehydes<sup>11a-c</sup> have been reported, asymmetric addition of acetylides to ketones to produce chiral tertiary alcohols is little known.<sup>11d,e</sup> A highly enantioselective addition of cyclopropanylacetylide to aryltrifluoromethyl ketone as a special substrate has been reported as the first example.11d,e However, a general method to prepare chiral tertiary  $\alpha$ -hydroxy acetylenes has not yet been reported.

Recently, we reported that chiral  $\alpha$ -keto amides derived from (S)-indoline-2-carboxylic acid resulted in high stereoselectivity in stereocontrolled additions of organometallics<sup>1b</sup> and allylation.<sup>1c</sup> On the supposition that these chiral  $\alpha$ -keto amides might be good chiral auxiliaries, we examined the diastereoselective additions of lithiated methoxyallene 2 and lithium acetylides 3 to various chiral  $\alpha$ -keto amides **1**.

The lithiated methoxyallene 2 was generated by treatment of methoxyallene (2.5 eq.) with *n*-BuLi (2.0 eq.) in THF at -78 °C for 30 min.<sup>7b</sup> Lithium acetylide **3** was prepared by addition of n-BuLi (1.5 eq.) to a solution of acetylene (1.7 eq.) in THF at 0 °C, followed by cooling to -78 °C after 30 min.<sup>11a</sup> Lithiated methoxyallene or lithium acetylide was reacted with  $\alpha$ -keto amides at -78 °C in THF. Since the allene adducts 4 are generally labile<sup>7b,c</sup> so giving low yields (Table 1), the crude product 4 was reacted to obtain 6 without purification. Two equivalents of 2 were added to 1 at -78 °C in THF. The reaction mixture was stirred for 10 min at -78 °C and then quenched with distilled water. Extraction with CH<sub>2</sub>Cl<sub>2</sub>, drying over  $MgSO_4$ , and concentration gave crude product 4, which was treated with a solution of t-BuOK (0.5 eq.) in DMSO at 50 °C. The reaction mixture was stirred for 1 h and then quenched with H<sub>2</sub>O. Extraction with Et<sub>2</sub>O, drying over MgSO<sub>4</sub>, and concentra-

† Electronic supplementary information (ESI) available: synthesis and spectroscopic data for 4b, 5g, 6b and 7b. See http://www.rsc.org/suppdata/ cc/b1/b100355k/

tion gave the crude residue, which was purified by column chromatography on  $SiO_2$  to give dihydrofuran derivatives 6 which were treated with 5% HCl and extracted with diethyl ether and EtOAc at pH 11. The combined organic layers were dried over MgSO<sub>4</sub> and concentrated to give dihydrofuranones 7. The results obtained are summarized in Table 2.

www.rsc.org/chemcomm

municatio

Table 1 Diastereoselective additions of lithiated methoxyallene 2 and lithium acetylides 3 to chiral  $\alpha$ -keto amides 1

		, <b>—</b> + ∟i—	Li 2 OMe or		OR'	or OR	OR' OH
Run	1 Substrate	R	3 R'	4 R″	Product	5 Yield (%) <sup>a</sup>	dr <sup>b</sup>
1	1a	Me	Me	_	4a	42	99:1
2	1b	Et	Me		4b	54	94:6
3	1c	<i>n</i> -Hex	Me		4c	45	98:2
4	1d	Ph	Me	_	4d	56	96:4
5	1e	Me	SiMe <sub>2</sub> Bu-t	_	4e	32	80:20
6	1f	Ph	SiMe <sub>2</sub> Bu-t		4f	33	83:17
7	1a	Me	Me	Ph	5a	71	98:2
8	1b	Et	Me	Ph	5b	91	94:6
9	1d	Ph	Me	Ph	5c	53	95:5
10	1e	Me	SiMe <sub>2</sub> Bu-t	Ph	5d	62	99:1
11	1a	Me	Me	CH <sub>2</sub> OMe	5e	72	99:1
12	1a	Me	Me	<i>n</i> -Bu	5f	88	99:1
13	1a	Me	Me	SiMe <sub>3</sub>	$5g^c$	66	> 99:1
<sup><i>a</i></sup> Isolated yields. <sup><i>b</i></sup> Determined by HPLC (Daicel chiral OD column). <sup><i>c</i></sup> $\mathbf{R}'' = \mathbf{H}$ : SiMe <sub>2</sub> was removed.							

Table 2 Transformation of crude products 4 to dihydrofuran derivatives 6 and 3(2H)-dihydrofuranones 7





Fig. 1 X-Ray structure of 5g

The purified products **4**, **5** and **6** were identified by <sup>1</sup>H, <sup>13</sup>C NMR,<sup>7b-d</sup> IR and MS. The ratios of diastereomers were determined by HPLC using a chiral OD column. The absolute configuration of **5g** was determined by comparison of the specific rotation of **8** ( $[\alpha]_D^{25} - 40.2^\circ, c = 1.7$ , acetone) with the literature value<sup>13</sup> ( $[\alpha]_D^{24} - 41.0^\circ, c = 10$ , acetone) and its structure determined by X-ray analysis (Fig. 1).<sup>14</sup> The ratios of diastereomers were unaltered during the process. Compound **8** has been found in plant growth regulators<sup>2</sup>*e* and highly enantiomeric excess synthesis of **8** has not been reported previously.

Enol ethers of 2,5-dihydrofuran derivatives 6 were readily hydrolyzed by HCl solution ( $H_2O:1,4$ -dioxane = 10:1) to provide the corresponding 3(2H)-dihydrofuranones 7 in good yields (75-80%), which are interesting intermediates as analogues of muscarone.<sup>15b</sup> Their structural element appears in other biologically active compounds<sup>15</sup> and their transformation to certain deoxy sugar derivatives can be performed. The indoline  $\alpha$ -keto amides have a great advantage in terms of cleavage of the amide bond to give chiral products and recover the indoline chiral auxiliary.<sup>1b,c,16</sup> The cleavage of the amide bond of indoline amides is much easier than that of alkyl amides such as proline amides. For instance, the chiral products 5g or 7b were readily hydrolyzed with 5% HCl in 1,4-dioxane under reflux for 3 h to give the corresponding 2-hydroxy-2-methylbut-3-ynoic acid 8 or 2-ethyl-3(2H)-dihydrofuranone-2-carboxylic acid 9 in 94–98% yields, respectively, as shown in Scheme 1. The chiral auxiliary was recovered in 95-98% yield without loss of optical purity. Earlier work<sup>11d,e</sup> on asymmetric addition of acetylide to a ketone (not a chiral auxiliary) gave one chiral tertiary alcohol from a specific ketone leading to one compound. Our method, however, provides a general methodology to produce chiral tertiary  $\alpha$ -hydroxy carboxylic acid acetylenes.





Conversion of the hydroxyalkylated allenes into the  $\alpha$ , $\beta$ unsaturated ketones under acidic conditions is also a useful reaction.<sup>9</sup> It is noteworthy that treatment of **4b** with 5% HCl provided enone **10** within 10 min at 25 °C as shown in Scheme 2. The ratio of diastereomers was also unaltered during hydrolysis. The enone moiety may be an interesting precursor for Michael-type additions or cycloadditions.<sup>17</sup>

In summary, it has been demonstrated that the reaction of  $\alpha$ keto amides derived from (*S*)-indoline-2-carboxylic acid with lithiated methoxyallene or lithium acetylide can provide useful



intermediates, chiral tertiary  $\alpha$ -hydroxy acid derivatives with high diastereoselectivities.

## Notes and references

- (a) K. Soai and M. Ishizaki, J. Org. Chem., 1986, **51**, 3290; (b) Y. H. Kim, I. S. Byun and J. Y. Choi, *Tetrahedron: Asymmetry*, 1995, **6**, 1025; (c) Y. H. Kim and S. H. Kim, *Tetrahedron Lett.*, 1995, **36**, 6895; (d) C. H. Senanayake, K. Fang, P. Grover, R. P. Bakale, C. P. Vandenbossch and S. A. Wald, *Tetrahedron Lett.*, 1999, **40**, 819, and references therein.
- 2 (a) S. Hanessian, Total Synthesis of Natural Products: The Chiron Approach, Pergamon Press, New York, 1983, ch. 2; (b) D. W. McPherson and F. F. Knapp, J. Org. Chem., 1996, **61**, 8335; (c) Y. Kanda and T. Fukuyama, J. Am. Chem. Soc., 1993, **115**, 8451; (d) A. V. R. Rao, J. S. Yadav and M. Valluri, Tetrahedron Lett., 1994, **35**, 3613; (e) W. L. White, K. L. Ricciardelli and M. K. Chaguturu, US Pat., 4704161.
- 3 T. Akiyama, K. Ishikawa and S. Ozaki, *Synlett*, 1994, 275; D. A. Evans, M. C. Kozlowski, S. C. Burgey and D. W. C. MacMillan, *J. Am. Chem. Soc.*, 1997, **119**, 7893; J. L. Wood, B. M. Stoltz, H. J. Dietrich, D. A. Pflum and D. T. Petsch, *J. Am. Chem. Soc.*, 1997, **119**, 9641; P. I. Dosa and G. C. Fu, *J. Am. Chem. Soc.*, 1998, **120**, 445.
- 4 F. J. Weiberth and S. S. Hall, J. Org. Chem., 1985, 50, 5308.
- 5 S. Hoff, L. Brandsma and J. F. Arens, *Recl. Trav. Chim. Pays-Bas*, 1968, **87**, 916.
- 6 For recent reviews of the chemistry of alkoxyallenes, see: R. Zimmer, *Synthesis*, 1993, 165; P. v. R. Schleyer, C. Lambert and E. U. Würthwein, *J. Org. Chem.*, 1993, **58**, 6377.
- 7 (a) D. Gange and P. Magnus, J. Am. Chem. Soc., 1978, 100, 7746; (b)
  S. W. Goldstein, L. E. Overman and M. H. Rabinowitz, J. Org. Chem., 1992, 57, 1179; (c) S. Hormuth and H.-U. Reissig, J. Org. Chem., 1994, 59, 67; (d) W. Schade and H.-U. Reissig, Synlett, 1999, 632.
- 8 P. Rochet, J.-M. Vatèle and J. Goré, *Synlett*, 1993, 105; T. Arnold, B. Orschel and H.-U. Reissig, *Angew. Chem., Int. Ed. Engl.*, 1992, **31**, 1033.
- 9 S. Hoff, L. Brandsma and J. F. Arens, *Recl. Trav. Chim. Pays-Bas*, 1968, **87**, 1179.
- 10 S. Hoff, L. Brandsma and J. F. Arens, *Recl. Trav. Chim. Pays-Bas*, 1969, 88, 609.
- (a) T. Mukaiyama and K. Suzuki, *Chem. Lett.*, 1980, 255; (b) G. M. R. Tombo, E. Didier and B. Loubinoux, *Synlett*, 1990, 547; (c) E. J. Corey and K. A. Cimprich, *J. Am. Chem. Soc.*, 1994, **116**, 3151; (d) A. S. Thompson, E. G. Corley, M. F. Huntington and E. J. J. Grabowski, *Tetrahedron Lett.*, 1995, **36**, 8937; (e) L. Tan, C. Chen, R. D. Tillyer, E. J. J. Grabowski and P. J. Reider, *Angew. Chem., Int. Ed.*, 1999, **38**, 711; (f) D. A. Evans, D. P. Halstead and B. D. Allison, *Tetrahedron Lett.*, 1999, **40**, 4461; (g) Z. Li, V. Upadhyay, A. E. DeCamp, L. Di Michele and P. J. Reider, *Synthesis*, 1999, 1453; (h) D. E. Frantz, R. Fässler and E. M. Carreira, *J. Am. Chem. Soc.*, 2000, **122**, 1806.
- K. C. Nicolaou and S. E. Webber, J. Am. Chem. Soc., 1984, 106, 5734;
   E. J. Corey, K. Niimura, Y. Konishi, S. Hashimoto and Y. Hamada, Tetrahedron Lett., 1986, 27, 2199; W. R. Roush and R. J. Sciotti, J. Am. Chem. Soc., 1994, 116, 6457; D. Vourloumis, K. D. Kim, J. L. Petersen and P. A. Magriotis, J. Org. Chem., 1996, 61, 4848.
- 13 D. Dugat, M. Verny and R. Vessière, Tetrahedron, 1971, 27, 1715.
- 14 Crystal data for **5g**:  $C_{15}H_{17}NO_3$ :  $M_r = 259.3$ , orthorhombic, space group  $P2_{12}_{12}_{12}$ , a = 10.300(2), b = 10.536(2), c = 25.810(6) Å, V = 2800.9(11) Å<sup>3</sup>, Z = 8,  $D_c = 1.230$  g cm<sup>-3</sup>, F(000) = 1104,  $\mu$ (Mo-K $\alpha$ ) = 0.086 mm<sup>-1</sup>,  $R_1 = 0.0333$ ,  $wR_2 = 0.0404$  for 2661 reflections ( $F_o > 4\sigma(F_o)$ ). (Note two independent molecules which have the same configuration are contained in the unit cell of the crystal of **5g**.) The absolute configuration could not be determined (Flack parameter: -0.43 (76)). CCDC 156841. See http://www.rsc.org/suppdata/cc/b1/b100355k/ for crystallographic data in .cif or other electronic format.
- 15 (a) J. E. Semple and M. M. Joullié, *Heterocycles*, 1980, 14, 1825; (b) M. DeAmici, C. Dallanoce, C. DeMicheli, E. Grana, A. Barbieri, H. Ladinsky, G. Schiavi and T. Zonta, J. Med. Chem., 1992, 35, 1915.
- 16 S. M. Kim, I. S. Byun and Y. H. Kim, Angew. Chem., Int. Ed., 2000, 39, 728.
- 17 For examples: W. Choy, L. A. Reed and S. Masamune, J. Org. Chem., 1983, 48, 1139.